

General Science 7th Class English Medium Chapter 6 Online Test

Sr	Questions	Answers Choice
1	Outermost shell of chlorine atom has..... electrons.	A. 1 B. 2 C. 6 D. 7
2	Oxygen is a	A. Monovalent element B. Divalent element C. Trivalent element D. Tertravalent element
3	Chemical bond in O ₂ is.	A. Ionic B. Covalent C. Metallic D. None of these
4	When water molecule gain a proton it form	A. H ⁺ B. OH ⁻ C. H ₃ O D. <pre style="margin: 0 0 0 40px; border: none; padding: 0px;">H2O</pre>
5	Types of covalent bond are	A. 1 B. 2 C. 3 D. 4
6	A molecule consists of two or more	A. Compounds B. Atoms C. Formulae D. Element
7	Valency of nitrogen is	A. 1 B. 2 C. 3 D. 4
8	Chemical bond that forms between metallic and non-metallic atoms.	A. Covalent B. Ionic C. Metallic D. None of these
9	Which of the following has double covalent bond.	A. Br ₂ B. CO ₂ C. HCl D. N ₂
10	While writing a chemical formula positive ions written on.	A. Right B. Left C. Center D. All of these
11	Negatively charged atom or group of atoms is called.	A. Proton B. Electron C. Anion D. Cation
12	Which of the following is cation.	A. H ₂ B. H ₂ O C. CO ₂ D. NH ₄ ⁺
13	An atom with stable electronic configuration can exist.	A. Reacts readily B. Can not exist C. Independently D. Combined form
14	Outermost shell of sodium atom has electrons.	A. 1 B. 2 C. 3 D. 4
15	Which of the following is an anion.	A. F B. Cl C. S-2

16 Chemical bond is O₂ is.

A. Ionic
B. Covalent
C. Metallic
D. None of these

17 Chemical bond between N₂ Molecule is

A. Ionic
B. Single covalent
C. Double covalent
D. Triple covalent

18 How many atoms form one molecule of chlorine.

A. 1
B. 2
C. 3
D. 4

19 Chemical formula of Sulphuric Acid.

A. HCl
B. H₂SO₄
C. HNO₃
D. CH₃COOH

20 Which of the following has stable electronic configuration.

A. H
B. He
C. C
D. Li
