

General Science 7th Class English Medium Chapter 6 Online Test

Sr	Questions	Answers Choice
1	Oxygen is a	A. Monoalent element B. Divalent element C. Trivalent element D. Tertravalent element
2	Which of the following has stable electronic configuration.	A. H B. He C. C D. Li
3	Chemical bond in O ₂ is.	A. Ionic B. Covalent C. Metallic D. None of these
4	While writing a chemical formula positive ion is written on.	A. Right B. Left C. Center D. All of these
5	Which of the following is a cation.	A. H ₂ B. H ₂ O C. CO ₂ D. NH ₄ ⁺
6	Chemical bond is formed between metallic and non-metallic atoms.	A. Covalent B. Ionic C. Metallic D. None of these
7	Which of the following has a double covalent bond.	A. Br ₂ B. CO ₂ C. HCl D. N ₂
8	Types of covalent bond are	A. 1 B. 2 C. 3 D. 4
9	Outermost shell of chlorine atom has..... electrons.	A. 1 B. 2 C. 6 D. 7
10	Which of the following is an anion.	A. F B. Cl C. S ²⁻ D. NH ₃
11	When a water molecule gains a proton it forms	A. H ⁺ B. OH ⁻ C. H ₃ O ⁺ D. <div style="border: 1px solid black; padding: 2px; display: inline-block;">H₂O</div>
12	Chemical bond in N ₂ molecule is	A. Ionic B. Single covalent C. Double covalent D. Triple covalent
13	Chemical formula of Sulphuric Acid.	A. HCl B. H ₂ SO ₄ C. HNO ₃ D. CH ₃ COOH
14	An atom with a stable electronic configuration can exist.	A. Reacts readily B. Can not exist C. Independently D. Combined form
15	A positively charged atom or group of atoms is called.	A. Anion B. Cation C. Neutron

		D. Proton
16	Outermost shell of sodium atom has electrons.	A. 1 B. 2 C. 3 D. 4
17	Valency of nitrogen is	A. 1 B. 2 C. 3 D. 4
18	How many atoms form one molecule of chlorine.	A. 1 B. 2 C. 3 D. 4
19	A molecule consists of two or more	A. Compounds B. Atoms C. Formulae D. Element
20	Negatively charged atom or group of atoms is called.	A. Proton B. Electron C. Anion D. Cation
21	Chemical bond in O ₂ is.	A. Ionic B. Covalent C. Metallic D. None of these