

## General Science 7th Class English Medium Chapter 6 Online Test

Sr	Questions	Answers Choice
1	Chemical bond is O <sub>2</sub> is.	A. Ionic B. Covalent C. Metallic D. None of these
2	Outermost shell of sodium atom has ..... electrons.	A. 1 B. 2 C. 3 D. 4
3	Which of the following has double covalent bond.	A. Br <sub>2</sub> B. CO <sub>2</sub> C. HCl D. N <sub>2</sub>
4	Valency of nitrogen is	A. 1 B. 2 C. 3 D. 4
5	A molecule consists of two or more	A. Compounds B. Atoms C. Formulae D. Element
6	Which of the following is an anion.	A. F B. Cl C. S <sup>2-</sup> D. NH <sub>3</sub>
7	When water molecule gain a proton it form	A. H <sup>+</sup> B. OH <sup>-</sup> C. H <sub>3</sub> O <sup>+</sup> D. <div style="border: 1px solid black; padding: 2px; display: inline-block;">H<sub>2</sub>O</div>
8	Outermost shell of chlorine atom has..... electrons.	A. 1 B. 2 C. 6 D. 7
9	Positively charged atom or grouping of atoms is called.	A. Anion B. Cation C. Neutron D. Proton
10	Types of covalent bond are	A. 1 B. 2 C. 3 D. 4
11	Which of the following has stable electronic configuration.	A. H B. He C. C D. Li
12	An atom with stable electronic configuration can exist.	A. Reacts readily B. Can not exist C. Independently D. Combined form
13	Negatively charged atom or group of atoms is called.	A. Proton B. Electron C. Anion D. Cation
14	Which of the following is cation.	A. H <sub>2</sub> B. H <sub>2</sub> O C. CO <sub>2</sub> D. NH <sub>4</sub> <sup>+</sup>
15	How many atoms form one molecule of chlorine.	A. 1 B. 2 C. 3 D. 4

		D. 4
16	Chemical bond teh forms between metallic and non-metallic atoms.	A. Covalent B. Ionic C. Metallic D. None of these
17	Oxygen is a	A. Monoalent element B. Divalent element C. Trivalent element D. Tertravalent element
18	Chemical bond in O2 is.	A. Ionic B. Covalent C. Metallic D. None of these
19	While writing a chemical formula positive ionis written on.	A. Right B. Left C. Center D. All of these
20	Chemical formula of Sulphuric Acid.	A. HCl B. H2SO4 C. HNO3 D. CH3COOH