

## General Science 7th Class English Medium Chapter 6 Online Test

Sr	Questions	Answers Choice
1	Outermost shell of chlorine atom has..... electrons.	A. 1 B. 2 C. 6 D. 7
2	Oxygen is a	A. Monovalent element B. Divalent element C. Trivalent element D. Tetravalent element
3	Chemical bond in O <sub>2</sub> is.	A. Ionic B. Covalent C. Metallic D. None of these
4	When water molecule gain a proton it form	A. H <sup>+</sup> B. OH <sup>-</sup> C. H <sub>3</sub> O <sup>+</sup> D. $\text{H}_2\text{O}$
5	Types of covalent bond are	A. 1 B. 2 C. 3 D. 4
6	A molecule consists of two or more	A. Compounds B. Atoms C. Formulae D. Element
7	Valency of nitrogen is	A. 1 B. 2 C. 3 D. 4
8	Chemical bond is formed between metallic and non-metallic atoms.	A. Covalent B. Ionic C. Metallic D. None of these
9	Which of the following has double covalent bond.	A. Br <sub>2</sub> B. CO <sub>2</sub> C. HCl D. N <sub>2</sub>
10	While writing a chemical formula positive ion is written on.	A. Right B. Left C. Center D. All of these
11	Negatively charged atom or group of atoms is called.	A. Proton B. Electron C. Anion D. Cation
12	Which of the following is cation.	A. H <sub>2</sub> B. H <sub>2</sub> O C. CO <sub>2</sub> D. NH <sub>4</sub> <sup>+</sup>
13	An atom with stable electronic configuration can exist.	A. Reacts readily B. Can not exist C. Independently D. Combined form
14	Outermost shell of sodium atom has ..... electrons.	A. 1 B. 2 C. 3 D. 4
15	Which of the following is an anion.	A. F B. Cl C. S <sup>2-</sup> D. ....

		D. $\text{NH}_3$
16	Chemical bond in $\text{O}_2$ is.	A. Ionic B. Covalent C. Metallic D. None of these
17	Chemical bond between $\text{N}_2$ molecule is	A. Ionic B. Single covalent C. Double covalent D. Triple covalent
18	How many atoms form one molecule of chlorine.	A. 1 B. 2 C. 3 D. 4
19	Chemical formula of Sulphuric Acid.	A. $\text{HCl}$ B. $\text{H}_2\text{SO}_4$ C. $\text{HNO}_3$ D. $\text{CH}_3\text{COOH}$
20	Which of the following has stable electronic configuration.	A. H B. He C. C D. Li