

## General Science 7th Class English Medium Chapter 4 Online Test

Sr	Questions	Answers Choice
1	The elements of group 18 are called.	A. Alkali metals B. Noble gases C. Alkaline metals D. Transition metals
2	The negatively charged particle around nucleus are called.	A. Electron B. Proton C. Atom D. Neutron
3	Atomic number of element present in period 2 and group 15 of the periodic table.	A. 4 B. 5 C. 6 D. 7
4	----- are fundamental particles of an atom.	A. Electron B. Proton C. Neutron D. All of these
5	Elements present in groups have same number of ..... in their outermost shell.	A. Electron B. Proton C. Atoms D. Neutron
6	The number of electrons present in an atom is equal to number of.	A. Nucleus B. Neutrons C. Proton D. Atom
7	The maximum number of electrons that N-Shell accommodate are.	A. 2 B. 8 C. 18 D. 32
8	Positively charged particle of an atom is called.	A. Electron B. Proton C. Atom D. Neutron
9	The charge present on atom is	A. Positive B. Negative C. Both a and b D. Neutral
10	The particles of an element having different mass numbers.	A. Proton B. Neutrons C. Isotopes D. Atoms
11	The particle of an atom having negative charge on it.	A. Electron B. Proton C. Neutron D. Nucleus
12	An element with fifteen protons in its atomic nucleus has atomic number.	A. 5 B. 10 C. 15 D. 20
13	The total number of electrons which M shell can accommodate	A. 2 B. 8 C. 18 D. 32
14	Area around nucleus where electrons are found are called.	A. Orbit B. Shell C. Both a and b D. None of these
15	The central part of an atom is called.	A. Electron B. Proton C. Neutron D. Nucleus

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16 The mass of an atom is almost due to the.  
A. Electron  
B. Proton  
**C. Nucleus**  
D. Neutrons

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17 How many periods are there in the Periodic table.  
A. Five  
B. Six  
**C. Seven**  
D. Eight

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18 Letter K is used to express the shell around the nucleus of an atom.  
A. 1st  
B. 2nd  
C. 3rd  
D. 4th

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19 Formula showing maximum number of electrons in a shell.  
A.  $n^2$   
**B.  $2n^2$**   
C.  $3n^2$   
D.  $4n^2$

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20 In .... elements are arranged in order of increasing atomic numbers.  
A. Group  
**B. Period**  
C. Orbit  
D. Shell

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